
Introduction

Tirana, Albania, 2 to 4 June 2010

Stephan Mieke

Introduction

- Measurement is present in almost every **human activity**, e.g. industrial, commercial, scientific, healthcare, safety and environmental.
- Measurement uncertainty is **associated** with any measurement.
- Measurement uncertainty plays a central role in **quality assessment** and quality standards.
- In **conformity assessment** the probability of making an incorrect decision based on the measurement and managing the consequential risks can be determined.
- As the **tolerances** applied in industrial production become more demanding, the role of measurement uncertainty becomes more important when assessing conformity to these tolerances.

Introduction

- **No measurement is exact.** When a quantity is measured, the outcome depends on the measuring system, the measurement procedure, the skill of the operator, the environment, and other effects.
- Even if the quantity were to be measured several times, **a different indication value** would in general be **obtained each time**.
- The **dispersion** of the indication values would relate to **how well the measurement is made**. Their **average** would **provide an estimate of the true quantity value** that generally would be **more reliable** than an individual indication value.
- It is not possible to state how well the true value of the measurand is known, but **only how well it is believed to be known**. Measurement uncertainty can be described as a measure of how well one believes one knows the true value of the measurand. This **uncertainty reflects the incomplete knowledge of the measurand**.

Introduction

Outline

- JCGM documents and other literature
- Traceability chain
- Definitions in the VIM
- Bayesian statistic
- Probability density functions

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Introduction - JCGM documents and other literature

Joint Committee for Guides in Metrology (JCGM):

- International Bureau of Weights and Measures (BIPM)
- International Electrotechnical Commission (IEC)
- International Federation of Clinical Chemistry and Laboratory Medicine (IFCC)
- International Laboratory Accreditation Cooperation (ILAC)
- International Organization for Standardization (ISO)
- International Union of Pure and Applied Chemistry (IUPAC)
- International Union of Pure and Applied Physics (IUPAP)
- International Organization of Legal Metrology (OIML)

Introduction - JCGM documents and other literature

Evaluation of measurement data

JCGM 100
Guide to the expression
of uncertainty in measurement
(**GUM**)

JCGM 104
An introduction to the GUM
and related documents

JCGM 101
Supplement 1 - Propagation
of distributions using a
Monte Carlo method

JCGM 102
Supplement 2 - Models with
any number of output values
(unpublished)

JCGM 103
Supplement 3 - Modelling
(unpublished)

Also on the agenda, but not published:

- JCGM 105 - Concepts and basic principles
- JCGM 106 - The role of measurement uncertainty in conformity assessment
- JCGM 107 - Applications of the least-squares method

Download from: <http://www.bipm.org/en/publications/guides/gum.html>

Introduction - JCGM documents and other literature

JCGM 100

Guide to the expression
of uncertainty in measurement
(GUM)

pros

- calculation by hand is possible
- in the framework of the Guide exact

cons

- linearization of the mathematical model required

JCGM 101

Supplement 1 - Propagation
of distributions using a
Monte Carlo method

pros

- no linearization of the mathematical model required
- shows the dispersion (PDF) of the result

cons

- requires software for practical application
- will give slightly different results every time the calculation is started

Introduction - JCGM documents and other literature

Other literature:

- The Expression of Uncertainty and Confidence in Measurement (UKAS M3003)

<http://www.ukas.com/library/Technical-Information/Pubs-Technical-Articles/Pubs-List/M3003.pdf>

- International vocabulary of metrology - Basic and general concepts and associated terms (VIM, 3rd ed., JCGM 200:2008)

<http://www.bipm.org/en/publications/guides/vim.html>

- Metrology in short, 3rd edition

<http://www.euramet.org/index.php?id=mis>

Introduction

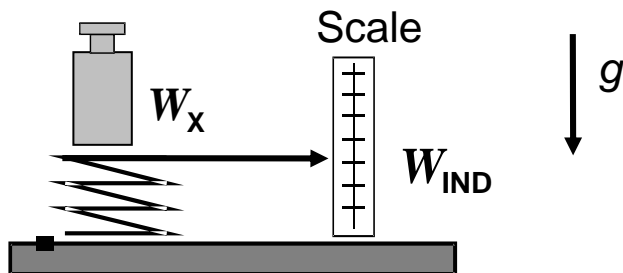
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Introduction - Traceability chain

Methods of measurement

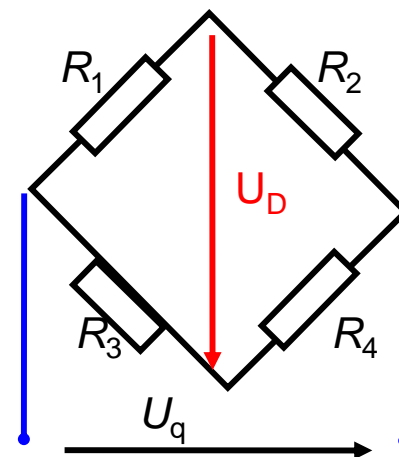
- Indication



The quantity (object) is used itself to generate the indicated reaction.

- Difference

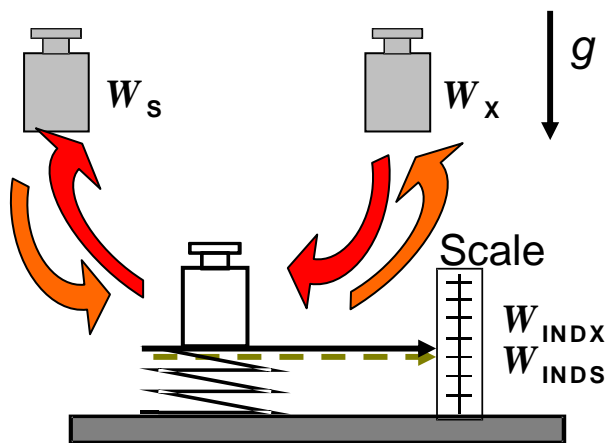
Example:
Wheatstone bridge



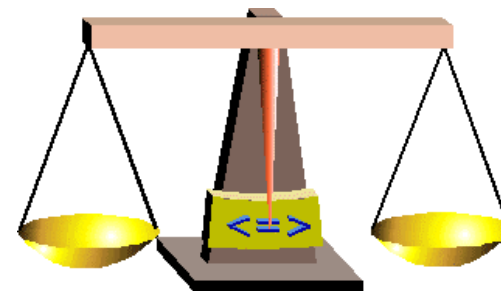
Determination of the difference of a quantity.

- Substitution

The value of quantity is determined using a standard and by taking alternate readings



- Compensation

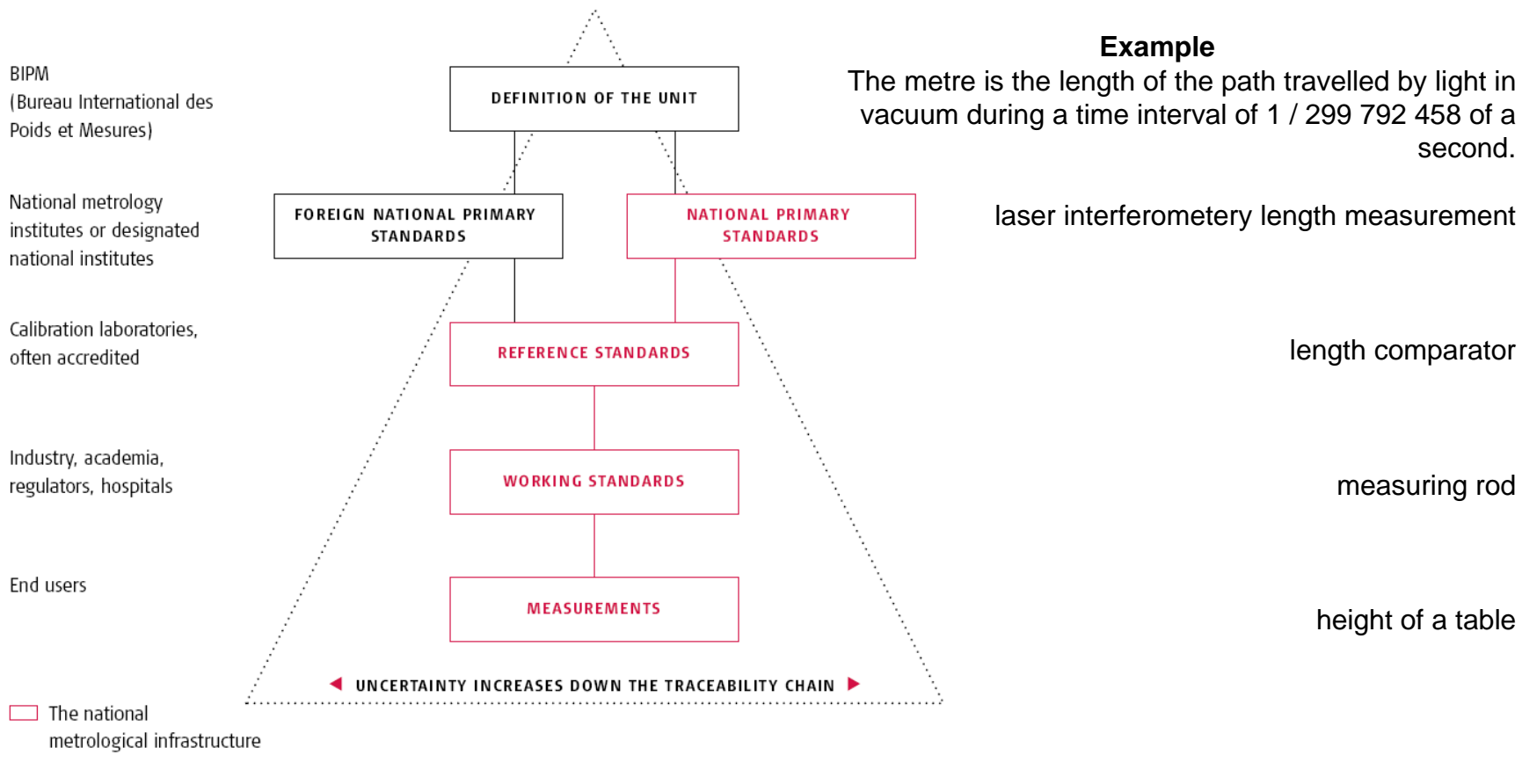


Simultaneous direct comparison (null method)

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Introduction - Traceability chain

Figure 1: The traceability chain



Traceability chain - the link to basic quantities

Expression of Measurement Uncertainty according to GUM

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Introduction - Definitions in the VIM

2.26 measurement uncertainty

non-negative parameter characterizing the dispersion of the **quantity values** being attributed to a **measurand**, based on the information used

Note: The definition in the GUM refers to the old VIM definition from 1993:

parameter, associated with the result of a measurement, that characterizes the dispersion of the values that could reasonably be attributed to the measurand

Introduction - Definitions in the VIM

1.1 quantity

property of a phenomenon, body, or substance, where the property has a magnitude that can be expressed as a number and a reference

1.19 quantity value

number and reference together expressing magnitude of a quantity

2.3 measurand

quantity intended to be measured

Introduction - Definitions in the VIM

1.1 quantity

property of a phenomenon, body, or substance, where the property has a magnitude that can be expressed as a number and a reference

length, l	radius, r	radius of circle A, r_A or $r(A)$
	wavelength, λ	wavelength of sodium D radiation, λ_D or $\lambda(D;Na)$
energy, E	kinetic energy, T	kinetic energy of particle i in a given system, T_i
	Heat, Q	heat of vaporization of sample i of water, Q_i
electric charge, Q		electric charge of the proton, e
electric resistance, R		electric resistance of resistor i in a given circuit, R_i
amount-of-substance concentration of entity B, c_B		amount-of-substance concentration of ethanol in wine sample i , $c_i(C_2H_5OH)$
number concentration of entity B, C_B		number concentration of erythrocytes in blood sample i , $C(Erys;B_i)$
Rockwell C hardness (150 kg load), $HRC(150\text{ kg})$		Rockwell C hardness of steel sample i , $HRC_i(150\text{ kg})$

Introduction - Definitions in the VIM

1.1 quantity

property of a phenomenon, body, or substance, where the property has a magnitude that can be expressed as a number and a reference

1.19 quantity value

number and reference together **expressing magnitude of a quantity**

EXAMPLE
Mass of a given body: 0,152 kg or 152 g

2.3 measurand

quantity intended to be measured

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- **Bayesian statistic**
- Probability density functions

Introduction - Bayesian statistic

Conventional statistic

The value of the measurand is assumed to be an unknown constant and the measurement data are random variables.

Bayesian statistic

The measurement data are constants and the value of the measurand is a random variable. The probability distribution for the value of the measurand describes the **degree of belief about all possible values** that could be attributed to the value of the measurand.

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Introduction - Probability density functions (PDF)

conventional statistic

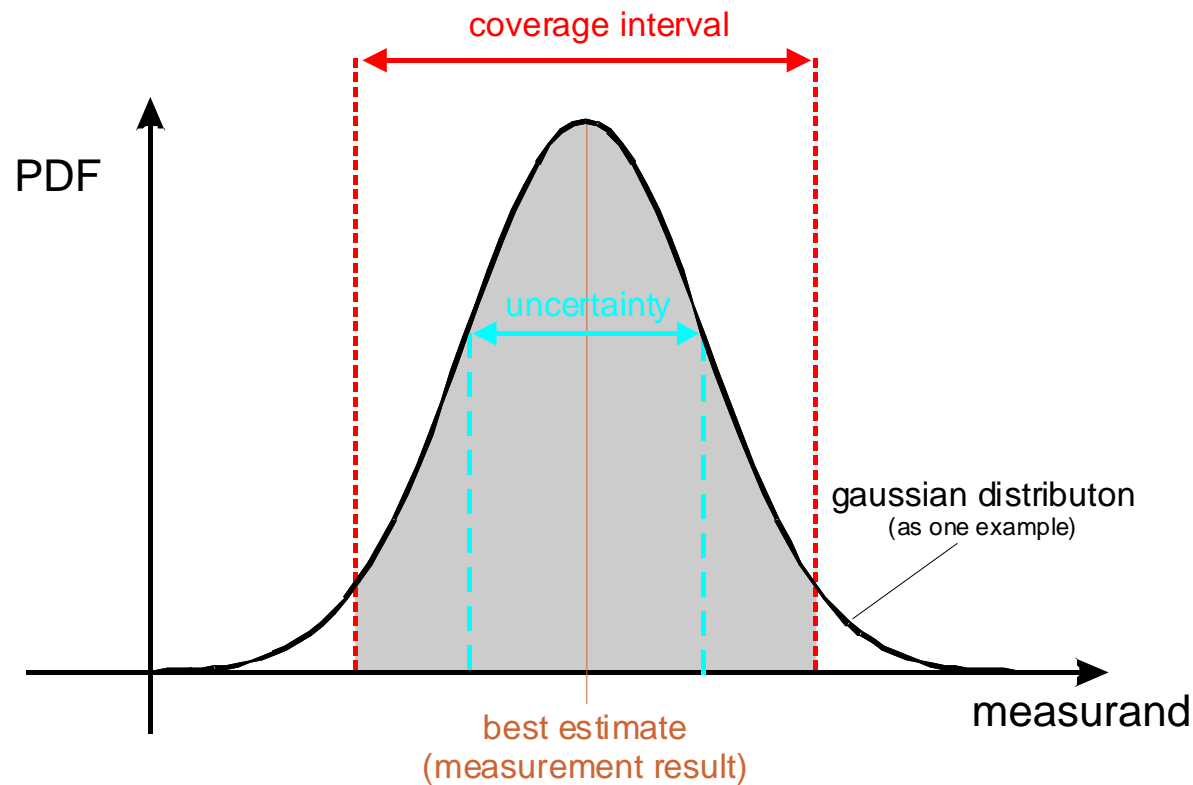
probability is understood to be the **relative frequency of the occurrence** of an observable random event in an infinite series of trials performed under identical conditions.

Bayesian statistic

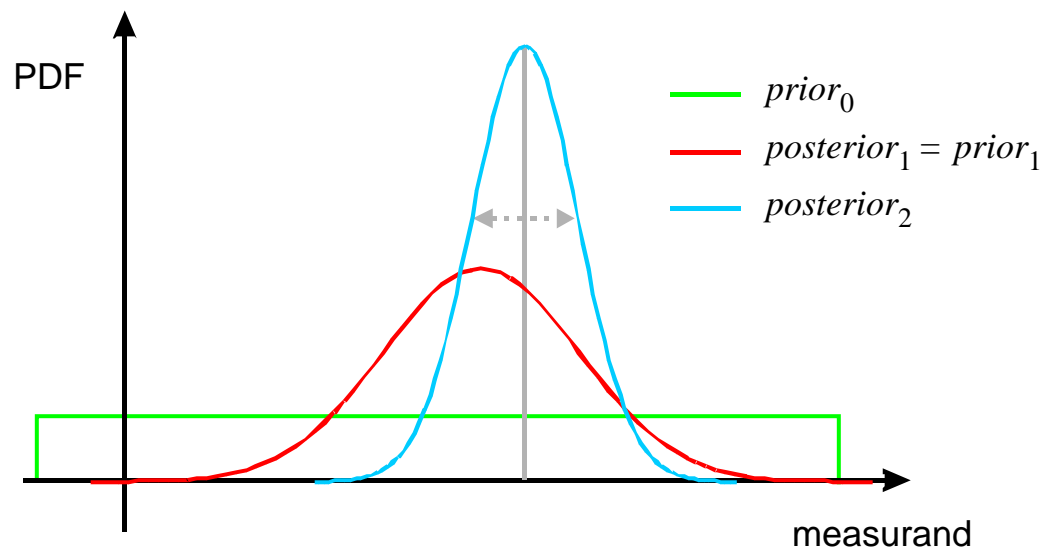
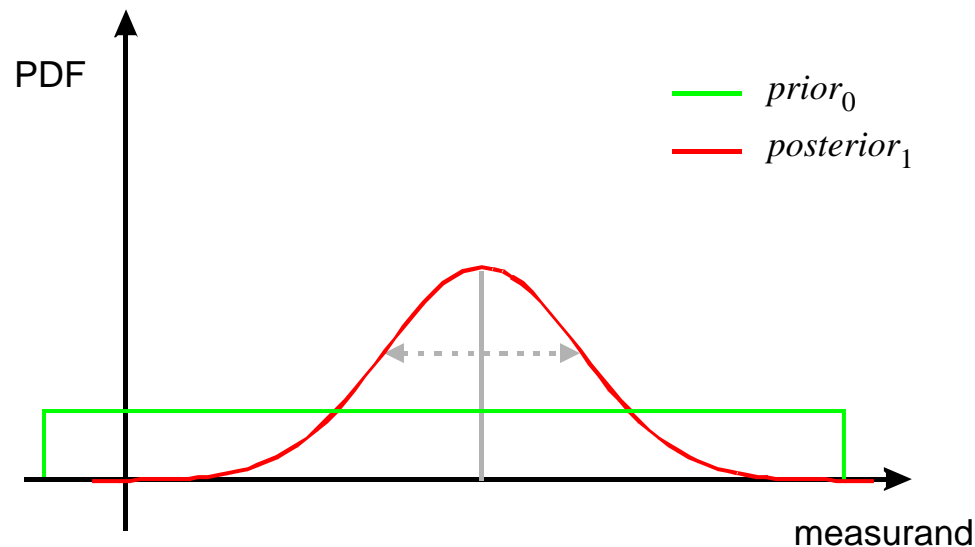
probability is understood to be a **degree of belief** or of the plausibility of a proposition, conditional on all relevant information that is available about that proposition.

Introduction - Probability density functions (PDF)

The probability density function describes the available knowledge about the quantity.



Introduction - Probability density functions (PDF)



more knowledge about the measurand

Introduction

Thank you for your attention

Questions ???

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